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# *Retrosynthetic Analysis of Peniapyrone E*

**Eric Wang**

*Jericho Senior High School, Jericho, USA*
*agedwy@gmail.com*

***Abstract.*** Retrosynthesis is the increasingly used strategy of breaking down a target molecule into simpler precursor structures through strategic disconnections. It is critical in helping devise a synthetic route to a complex organic compound as it ideally fragments the original molecule into easily obtainable pieces. The goal of this paper is to provide such a plan that would open a simple synthetic pathway to the recently discovered organic molecule Penipyrone E.

***Keywords:*** Retrosynthesis, Retrosynthetic analysis, Peniapyrone E

## 1. Introduction

The synthesis of complex natural products remains one of the most challenging yet rewarding endeavors in organic chemistry. Discovering a simpler way to synthesize a globally used compound such as Penicillin V or morphine would save the drug industry millions of dollars, while making these compounds more accessible to everyone who needs them.

Peniapyrone E was one of 9 “Five cyclopenta[d]pyrano[4,3-b]pyran-1,7(6H)-dione 6/6/5-fused tricyclic ring system containing metabolites” isolated from the fungus Penicillium brefeldianum [1]. Peniapyrone E, among others, showed “cytotoxic activity against AsPC-1, CRL-2234, and MCF-7 cancer cell lines”, giving reason for a simplistic synthesis pathway to be developed for it [2].

The retrosynthetic steps in this paper are taken after careful consideration of oxidation states, dioxygenation patterns, and the resulting products(or rather, potential reactants) [3-5].

## 2. Retrosynthetic analysis

To begin the retrosynthetic process, we first examine the target molecule, Peniapyrone E, shown in Figure 1.

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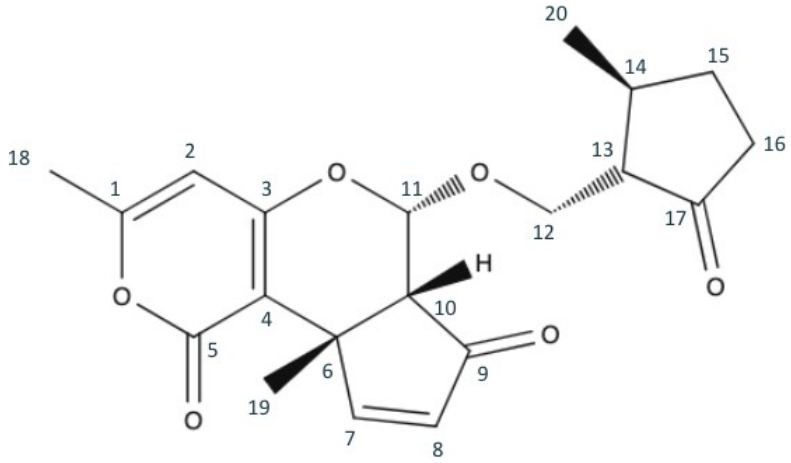


Figure 1. Peniapyrone E

Upon first observation, we can easily see that Peniapyrone E consists of two main parts: the complex triple-ring structure to the left, and the 3,methylcyclopentanone structure to the right. The logical retrosynthetic step is to separate these structures from each other, and it is best done by breaking the hash wedged C-11 bond. Since C-11 was originally in the aldehyde oxidation level, its bond could be broken into two alcohols with an aldehyde left over. It will be simple for these pieces to be synthetically reattached. The resulting products are seen in figure 2.

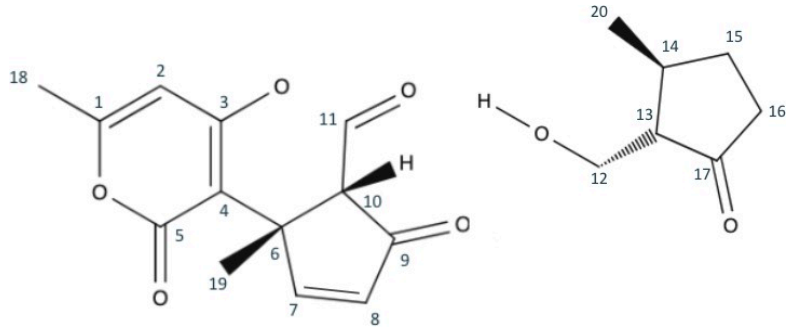


Figure 2. Peniapyrone E after a C-11 bond disconnection

Continuing the retrosynthesis, we shift our attention to the cyclopentane structure to the right. Taking advantage of the 1,3 dioxygenation pattern in carbons 12, 13, and 17, we can break bond C12-C13. By disconnecting this bond, the alcohol tail is turned into a formaldehyde and the cyclopentane becomes 3,methylcyclopentanone with an anion at carbon 13(shown in Figure 3). A synthetic equivalent would be required to represent the anion, but formaldehyde already has a partial positive charge at its carbon.

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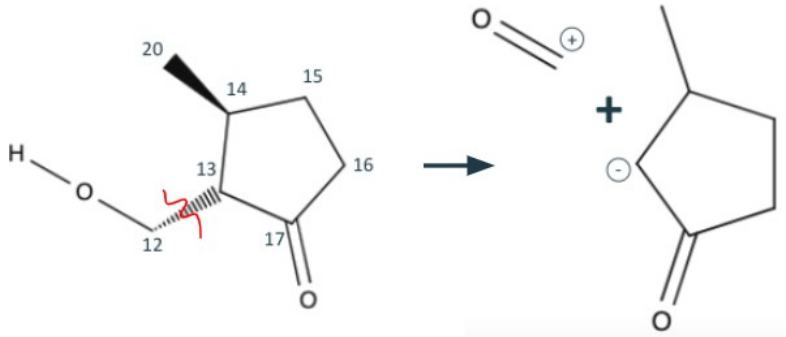

Figure 3. Disconnection of bond C12-C13

Now we continue to retrosynthesize what remains of the triple-ring system (on the left of Figure 2). It now consists of a benzene-like and a cyclopentene structure, connected by bond

C4-C6. We could disconnect this bond right here, but the resulting methylcyclopentenedione will be a difficult molecule to simplify. Instead, we can opt to do some fine tuning instead. First, we do a functional group removal and remove the olefin from the cyclopentene structure(Figure 4).

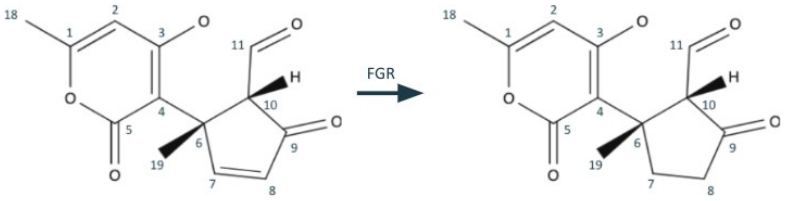


Figure 4. FGR of olefin C7=C8

In a synthesis, this would be accommodated by a desaturation. Now, we can utilize the 1,5 dioxygenation pattern at C5, C4, C6, C10, and C11 to justify the C4-C6 disconnection, shown in Figure 5.

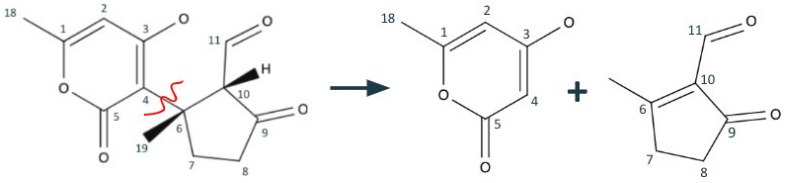

Figure 5. Disconnection of bond C4-C6

The rightward cyclopentane molecule might look a bit familiar. This is because it's almost identical to the cyclopentane in figure 2. Having two identical molecules to work for at different stages of a synthesis would be very helpful in reducing the complexity of the synthesis process.

Therefore, using functional group interconversion, we can convert the aldehyde at C11 into an alcohol to take advantage of the potential 1,3 dioxygenation pattern at C9-11. Then, we use functional group removal to take care of the olefin, similar to what we did for C7=C8.

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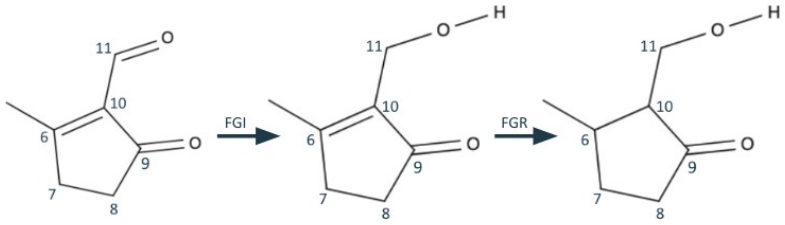


Figure 6. Fine tuning of C6-11 structure

Finally, we use the resulting 1,3 dioxygenation pattern to break C10-C11, much like we broke C12-C13. We end with the synthons identical to those in Figure 3: formaldehyde and a 3,methylcyclopentanone enolate.

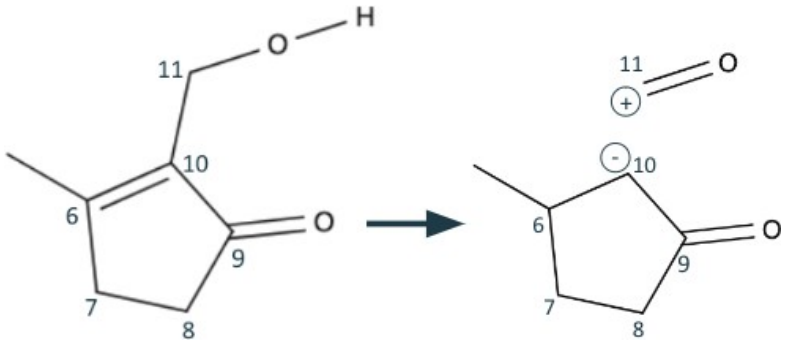

Figure 7. Disconnection of bond C10-C11

The last steps of this retrosynthetic process involves the quasi-benzene structure in the middle of Figure 5. Although a retrosynthetic step might not immediately be apparent, carbons 3-5 are very close to being a 1,3 dioxygenation pattern. Therefore, we do two functional group interconversions on the neighboring bonds of carbon 3. The first is a hydrogen shift from C3-O to C3=C3, removing the olefin and forming an aldehyde. The second is an interconversion from the resulting aldehyde into an alcohol.

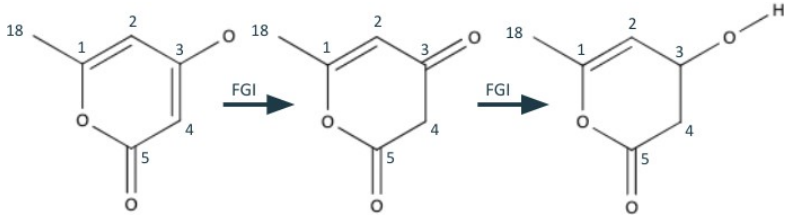


Figure 8. Fine tuning of C1-5, C18 structure

Finally, we can use the newly formed 1,3 dioxygenation pattern to break bond C3-C4 and turn the 6-membered ring into a carbon and oxygen chain.

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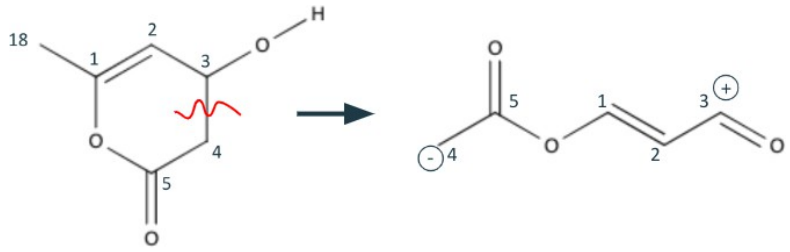


Figure 9. Disconnection of bond C3-C4

## 3. Conclusion

This paper presents a detailed retrosynthetic analysis of Peniapyrone E, a complex natural product with promising cytotoxic activity. The process of retrosynthesis involved careful consideration of oxidation states and dioxygenation patterns, leading to a series of manageable synthetic intermediates. These results should provide a framework that would aid in a possible synthetic process of Peniapyrone E. For example, many retrosynthetic steps involving a 1,3 dioxygenation pattern could be reversed with an aldol addition reaction, while 1,5 disconnections could be reversed with a michael addition process [2]. As shown in this paper, retrosynthetic analysis is a technique that will help many synthetic chemists devise a synthetic pathway to their target molecules.

## References

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